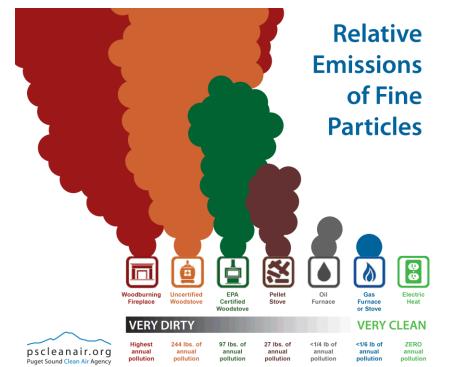


Interrelated Scientific Principles: Matter, Energy, and Environment

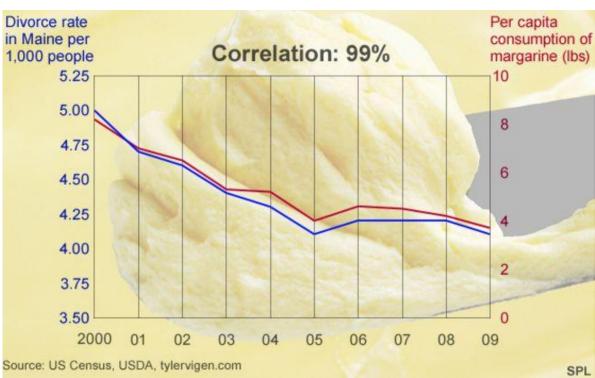
WOOD STOVES = AIR POLLUTION

- EPA estimates wood stoves are responsible for 5% of the smallest, deadliest particles emitted in the US.
- •¾ of current stoves are pre-EPA certified
- EPA education campaign using fee from sale of wood-burning device



SCIENCE

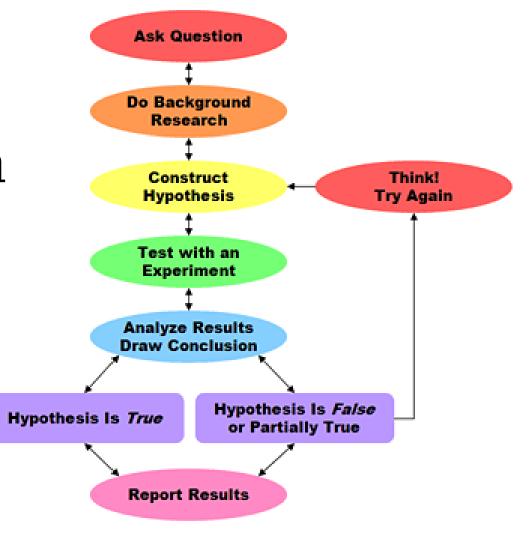
- Process to solve problems or understand nature that involves testing possible answers.
- <u>Cause and effect relationship</u>: Winter rain leads to spring "green-up"
- Not every correlation implies a cause and effect:





CONTROLLED EXPERIMENTS

When two groups differ in Only one way
High sample size
Reproducible results



SCIENTIFIC METHOD IN ACTION -ACID RAIN

- Fish and forests dying
- Tested pH of rain and lakes



- Tested emissions from upwind factories
- Tested pH of rivers upwind and downwind of factories
- Reduced emissions and re-tested



"FRQ STEW"

- 1) Reads FRQ question aloud
- 2) Gives answer
- 3) Gives answer (states whether agrees or disagrees why?/why not?)
- 4) Approve/Correct using Scoring Guide
- Rotate roles after each question
- Everyone write answers and corrections



DESIGN AN EXPERIMENT WITH YOUR GROUP

- A student wants to test the effects of caffeine on the growth rate of sunflowers.
- Write a hypothesis
- Design an experiment to test this hypothesis.
- •What is the control?
- •What is the dependent variable?
- •How will you know if the hypothesis is supported?





WHICH IS THE CONTROL?

•A. The group that receives the blood pressure medicine.

B. The group that receives the placebo



WHY IS A CONTROL NECESSARY?



WHICH WOULD GIVE THE MOST RELIABLE RESULTS?

A) A sample size of 10
B) A sample size of 100
C) A sample size of 1000



WHICH IS THE DEPENDENT VARIABLE?

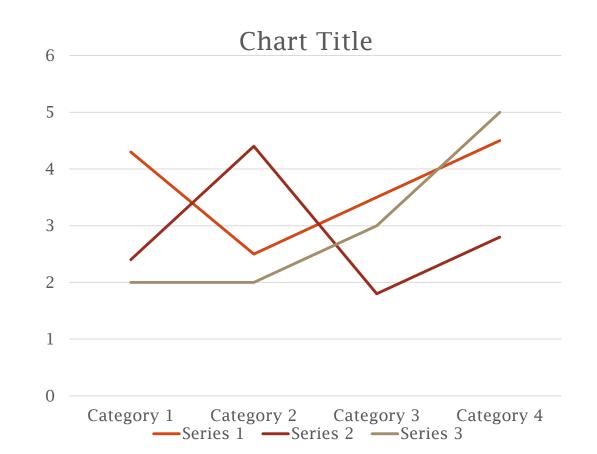
A. The blood pressure of the individualsB. The type and dose of medication givenC. The timing of the medication



WHICH AXIS OF A GRAPH WOULD THE DEPENDENT VARIABLE GO ON?

B) Y Axis

A) X Axis





THEORIES AND LAWS

- Laws describe what happens and
- theories describe why things happen



- Law: Uniform, constant fact in nature
- <u>Theory</u>: widely accepted generalization that explains why things happen
 - Supported by massive amount of evidence/data
 - Broad concept that shapes how scientists look at the world
 - Germ theory, molecular theory, gravitational theory, evolutionary theory



"Just a Theory?"

Myth: In science if it's not called a law it's not proven, thus it's just a guess.

Observable Fact	Current Best Explanation of the Fact
Gravity	Theory of Relativity and Special Relativity (A. Einstein)
Infectious Disease	Germ Theory of Disease (A. Bassi)
Evolution	Theory of Evolution by Natural Selection* (C.
Earthquakes and Volcanoes	Theory of Plate Tectonics (J. Tuzo Wilson)



PSEUDOSCIENCE

 A deceptive practice that uses the appearance or language of science to convince, confuse or mislead.

Not supported by unbiased tests or evidence





HAND-UP, STAND-UP, PAIR-UP!

- 1. "<u>Rally-Robin</u>" pseudoscientific ideas (back and forth)
- 2. Switch partners (when instructed) and <u>repeat</u>
- 3. Share-out



SHARE-OUT

- Many infomercials like alkaline water
- Astrology
- Paranormal investigators
- Big Foot, mermaids etc.









Science	Pseudoscience
Willingness to change with new evidence	Fixed ideas
Ruthless peer review	No peer review
Takes account of all new discoveries	Selects only favourable discoveries
Invites criticism	Sees criticism as conspiracy
Verifiable results	Non-repeatable results
Limits claims of usefulness	Claims of widespread usefulness
Accurate measurement	"Ball-park" measurement

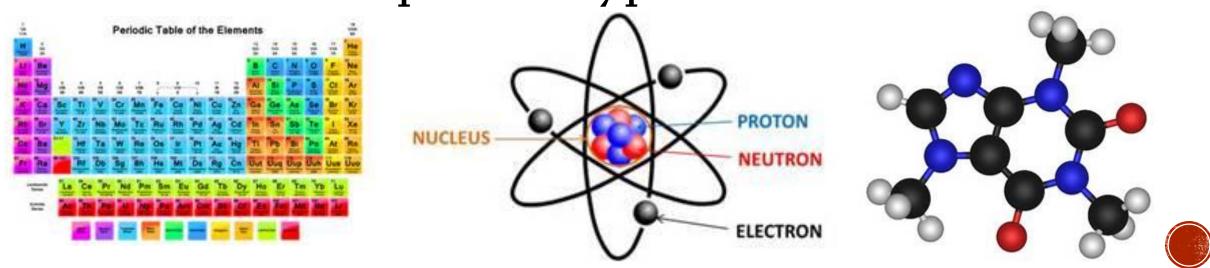






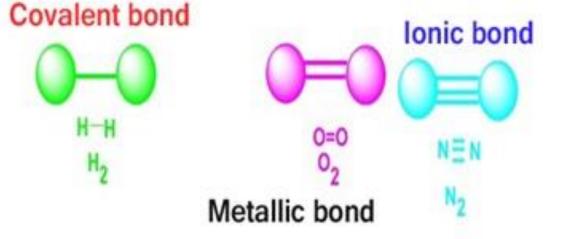
STRUCTURE OF MATTER

- Matter is anything that takes up space and has mass
- Atoms are fundamental units of matterElements are specific types of matter



CHEMICAL BONDS

- Physical attractions between atoms resulting from the interaction of their electrons.
- When bonds are broken or formed, a chemical reaction occurs, and the amount of energy is changed.



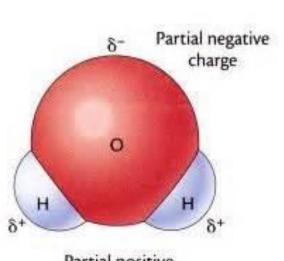






Polar (negative on one end, positive on the other)

- Cohesive



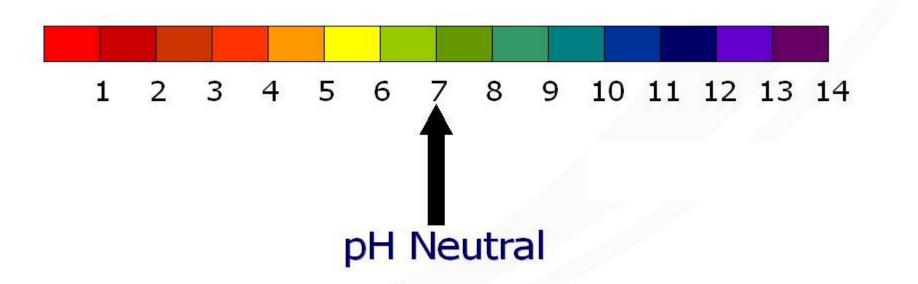
- Evaporation of water cools surroundings
- Universal solvent



Partial positive charge

ACIDS AND BASES

- <u>Acid</u>: releases hydrogen ions (protons) in solution (below pH of 7) - acidic
- <u>Base</u>: accepts hydrogen ions in solution (above pH of 7) alkaline





GET YOUR WHITEBOARDS READY!!!





CHALLENGE #1:

Write the equation for photosynthesis



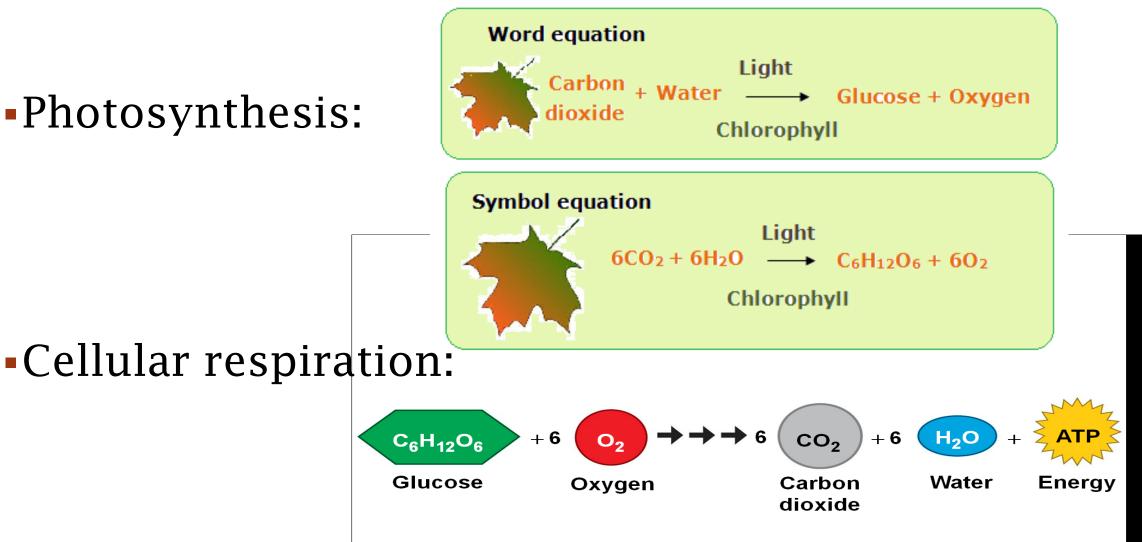
CHALLENGE #2:

Write the equation for cellular respiration



CHEMICAL REACTIONS IN LIVING THINGS **Photosynthesis**

Photosynthesis:





ENERGY IS THE ABILITY TO DO WORK



 <u>Kinetic energy</u>: energy contained by moving objects (moving air molecules, water in a river etc.)

 <u>Potential energy</u>: dependent on position (ex. Plants, water behind a dam)





THERMODYNAMICS

- First law of thermodynamics: energy can neither be created nor destroyed; it can only be changed from one form to another.
- Second law of thermodynamics: whenever energy is converted from one from to another, some of the useful energy is lost.
 Entropy: energy that cannot be used to do useful work



ENVIRONMENTAL IMPLICATIONS OF ENERGY FLOW

Entropy increases: houses decompose,

automobiles rust etc. (releases heat)



Energy quality:

high quality can be harnessed and used (ex. electrical)



 Low quality can still have significance – ocean temperatures affecting weather patterns etc. and can sometimes be used with the right technology (wind turbines etc.)



DIESEL ENGINE TRADE-OFFS



- Diesel engine efficiency: 35-42% (more particulate matter and nitrogen oxides)
- Gasoline engine efficiency: 25-30% (20-40% more carbon dioxide)

•Should the US switch to diesel engines like Europe?

